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ChemFacts. • The primary source of hydrocarbons is petroleum.  
• About 75 million barrels of petroleum are pumped out of the

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Earth each day. • Hydrocarbons are used as fuels and are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

## **Chapter 21: Hydrocarbons**

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...

Chemistry is the study of matter: its composition, properties, and reactivity. This material roughly covers a first-year high school or college course, and a good understanding of algebra is helpful.

Did you know that everything is made out of chemicals?

Chemistry is the study of matter: its composition, properties, and reactivity.

## **Chemistry | Science | Khan Academy**

A more appropriate definition of organic chemistry is: The chemistry of hydrocarbons (compounds of carbon and hydrogen only) and the derivatives of hydrocarbons, (compounds containing C, H, and O, N, S, P, halogens etc). In some organic compounds, the metals such as Na, K, Ca and many others can be found.

## **A SELF-STUDY GUIDE TO THE PRINCIPLES OF ORGANIC CHEMISTRY**

22.1 Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules.

## **CHAPTER 22 Study Guide - Quia**

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Answers rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane Study Guide: Chapter 22 Hydrocarbon Compounds Flashcards... 22.1 Hydrocarbons.

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## Chapter 22 Hydrocarbons Study Guide Answers

HYDROCARBONS - Qld Science Teachers Hydrocarbons with double or triple bonds are called unsaturated In alkenes, a double bond is one in which two pairs of electrons are shared between carbon atoms For example, the smallest alkene is ethene with the chemical formula of  $C_2H_2$  and the

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Question 48 Identify the aromatic hydrocarbon with the highest stabilization energy. phenanthrene benzene naphthalene anthracene By accessing this Question Assistance, you will learn while you earn points based on the Point Pot Question Att

### Solved: Question 48 Identify The Aromatic Hydrocarbon With ...

SCIENCE A CHEMISTRY Answer all questions 2 This question is about hydrocarbons 2 (a) Most of the hydrocarbons in crude oil are alkanes 2 (a) (i) Large alkane molecules can be cracked to produce more useful molecules 3 Metals are extracted from ores in the Earth's crust

### [MOBI] Hydrocarbons From Earths Crust Answer Key

STUDY GUIDE In your textbook, read about organic chemistry, hydrocarbons, and straight-chain alkanes. Use each of the terms below just once to complete the passage. homologous series straight-chain alkanes If all of the carbon atoms are linked by single covalent bonds and there are no branches, the compounds are called (1) examples.

### KMBT 654-20150625115832

Benzene,  $C_6H_6$ , is the simplest member of a large family of hydrocarbons, called aromatic hydrocarbons. These compounds contain ring structures and exhibit bonding that must be described using the resonance hybrid concept of valence bond theory or the delocalization concept of molecular orbital theory.

### 21.7: Aromatic Hydrocarbons - Chemistry LibreTexts

Hydrocarbons. Alkanes. Alkenes. the study of carbon and its compounds. simplest organic compound, composed of hydrogen and carbon. hydrocarbons with single bonds, "ane" ending.

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hydrocarbons with double bonds, "ene" ending. Organic Chemistry. the study of carbon and its compounds.

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