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Review Solutions Section 3

Answers

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### Answers

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### Answers

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Answers

## **Chapter 12 Review Solutions Section**

Modern Chemistry 5 Solutions CHAPTER  
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ANSWER Answer the following questions  
in the space provided. 1. The following  
are statements about the dissolving  
process. Explain each one at the  
molecular level. a. Increasing the

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## Answers

pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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Modern Chemistry 105 Solutions  
CHAPTER 12 REVIEW Solutions SECTION  
2 SHORT ANSWER Answer the following  
questions in the space provided. 1. The  
following are statements about the

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## Answers

dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the

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CHAPTER 12 REVIEW Solutions SECTION  
1 SHORT ANSWER Answer the following

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## Review Solutions Section 3

### Answers

questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:



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Answers

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## Answers

they know. They answer questions comparing solutions, suspensions, and colloids.

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## Answers

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## Answers

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## Answers

Chapter 12 Review Solutions Section  
Modern Chemistry 7 Solutions CHAPTER  
12 REVIEW Solutions SECTION 3 SHORT  
ANSWER Answer the following questions  
in the space provided. 1. Describe the  
errors made by the following students in  
making molar solutions. a. James needs  
a 0.600 M solution of KCl. He measures  
out 0.600 g of KCl and adds 1 L of water

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## **Chapter 12 Review Solutions Section 3 - mail.trempealeau.net**

Modern Chemistry Chapter 12-Solutions  
Section 1- Types of Mixtures Solutions  
are homogeneous mixtures of two or  
more substances in a single phase.  
Soluble describes a substance as

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capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

## **Modern Chemistry Chapter 12 Section 1 Review Answers**

CHAPTER . 12 . REVIEW ""-- Solutions .  
SHORT ANSWER Answer the following

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## Answers

questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

## **CHAPTER REVIEW Solutions**



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## Answers

Modern Chemistry Chapter 12-Solutions  
Section 1- Types of Mixtures Solutions  
are homogeneous mixtures of two or  
more substances in a single phase.  
Soluble describes a substance as  
capable of being dissolved. Solvent is  
the dissolving medium in a solution.  
Solute is the substance that is dissolved  
in a solution.

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question, ONLY WRITE THE PROBLEM  
NUMBER and ANSWERS/SOLUTIONS. •  
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## Answers

you must show your work to get full credit. • For multiple choice questions, you can simply write the letter (a, b, c, or d) of the correct response.

### **Chapter 12 Review Solutions Section 1 Answer Key**

Modern Chemistry 103 Solutions

CHAPTER 12 REVIEW Solutions SECTION

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## Answers

1 SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. \_\_\_\_\_ solutions (a) larger than 1000 nm \_\_\_\_\_ suspensions (b) 1 nm to 1000 nm

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Answers

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## Answers

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1 SHORT ANSWER 1. c 2. a 3. b 2. a.

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## Answers

alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text.

## **Chemistry Chapter 12 Section 1 Answers**

Modern Chemistry Chapter 12-.  
Solutions. Section 1- Types of Mixtures



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## Answers

Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

### **Modern Chemistry Chapter 12- Solutions**

Chemistry Chapter 12 section 2. solution

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## Review Solutions Section 3

### Answers

equilibrium. saturated solution.  
unsaturated solution. supersaturated  
solution. the physical state in which the  
opposing process of dissolutio.... a  
solution that cannot dissolve any more  
solute under the give.... A solution that  
contains less solute than a saturated  
solution....

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